

# Peculiarities of Lattice Dynamics and the Melting of Alkali Metals

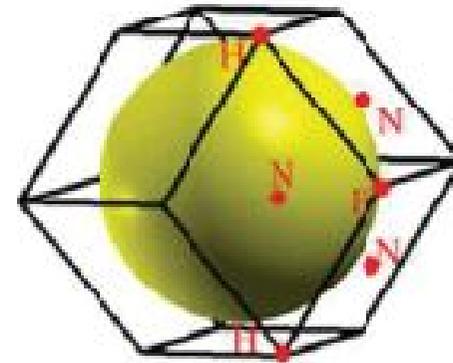
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1. Introduction  
*(from simplicity to complexity)*
2. Melting of sodium  
*(simplicity including complexity)*
3. Melting of lithium  
*(near total complexity)*
4. Conclusions  
*(what is in the future?)*

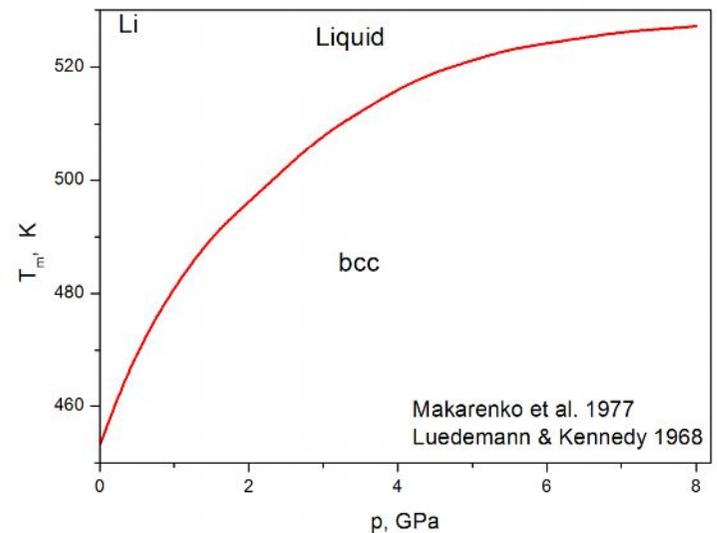
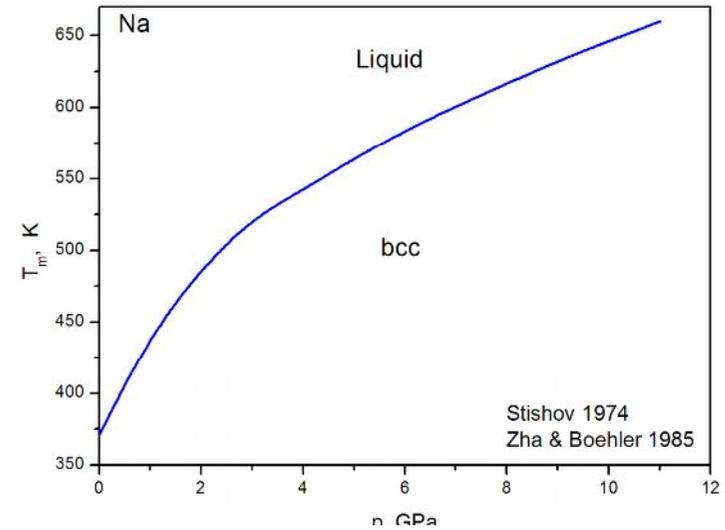
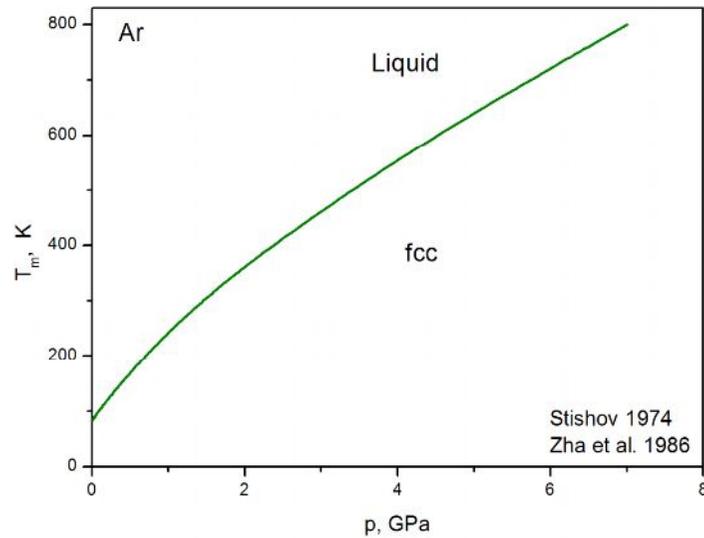
*S.V. Lepeshkin et al. JETP Lett. (2009)*

# At ambient conditions, the alkali elements are prototype free-electron metals:

- highly symmetric crystal structure
- nearly parabolic  $E(\mathbf{k})$
- spherical Fermi surface
- small electron-phonon interaction → no superconductivity
- simple thermodynamic properties



# Experimental melting curves of Ar, Na, and Li



Melting in classical systems is well described by Lindemann's criterion:

$$\sqrt{\langle u^2(T_m) \rangle} = Ld_{NN}$$

with  $L \approx \text{const}$

## Structural sequences for alkali elements at $T \sim 300$ K

Li	bcc $\underline{7.5}$ > fcc $\underline{39}$ > hR1 $\underline{42}$ > cI16 $\underline{69}$ > Li-6 $\underline{79}$ > <b>Li-7</b> < 110 GPa 9R, fcc (low $T$ )
Na	bcc $\underline{65}$ > fcc $\underline{105}$ > cI16 $\underline{118}$ > oP8 $\underline{125}$ > host-guest $\underline{200}$ > <b>hP4</b> < 220 GPa
K	bcc $\underline{11.6}$ > fcc $\underline{20}$ > host-guest < 60 GPa
Rb	bcc $\underline{7}$ > fcc $\underline{13}$ > oC52 $\underline{17}$ > host-guest $\underline{20}$ > tI4 $\underline{48}$ > oC16 < 70 GPa
Cs	bcc $\underline{2.4}$ > fcc $\underline{4.2}$ > oC84 $\underline{4.4}$ > tI4 $\underline{12}$ > oC16 $\underline{72}$ > dhcp < 223 GPa <b>Cs becomes superconductive in the fcc phase. At <math>p \sim 12</math> GPa <math>T_c \sim 1.5</math> K</b>

**fcc Li becomes superconductive at  $p > 20$  GPa.  $T_c \sim 14$  K at 33 GPa**

**At  $p \sim 80$  GPa, Li-7 becomes semiconducting with  $E_g \sim 0.1$  eV**

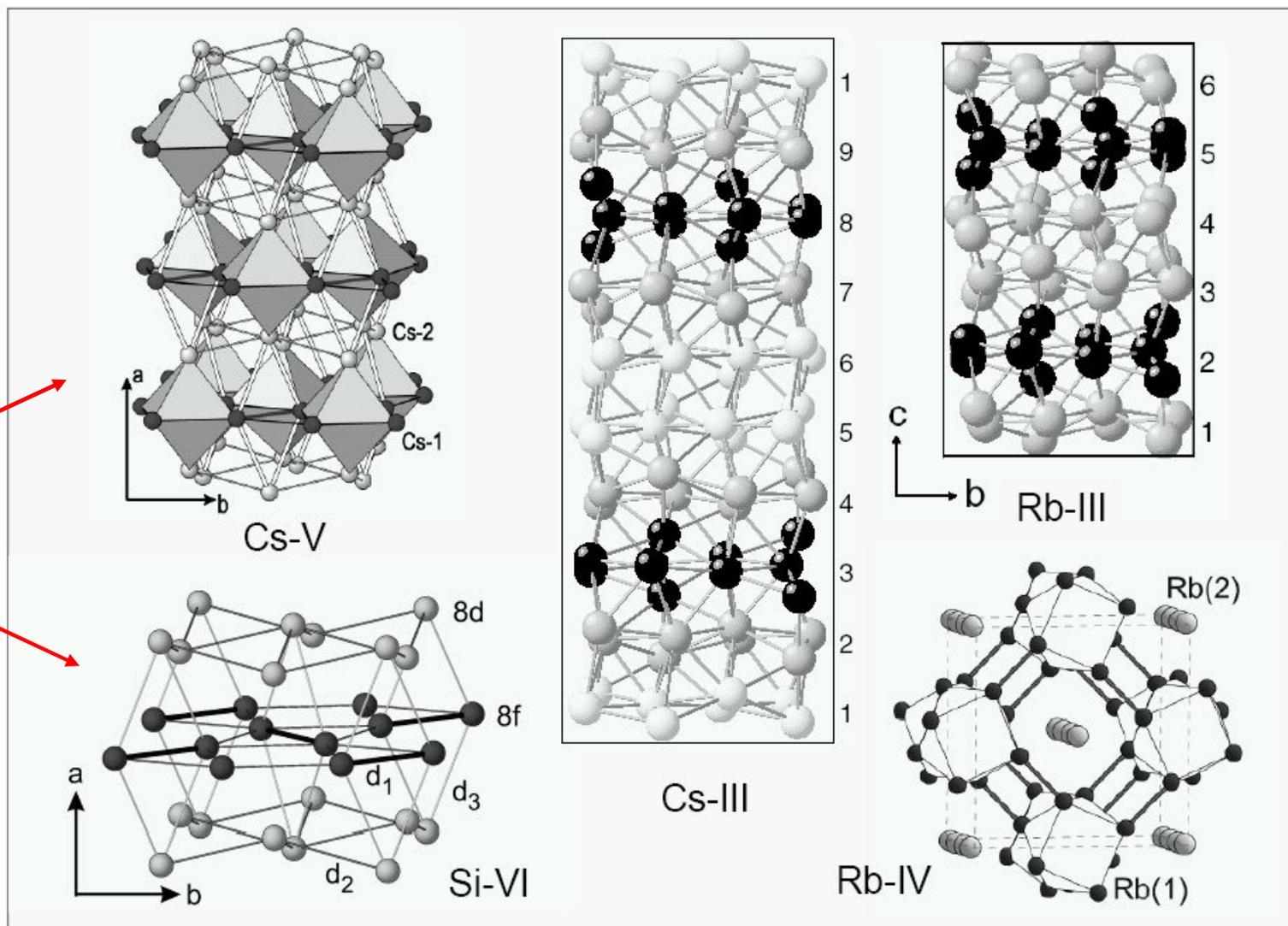
**At  $p \sim 200$  GPa, Na is in the semiconducting phase hP4 ( $E_g \sim 2$  eV)**

# High pressure phases of alkali metals

*oC84*

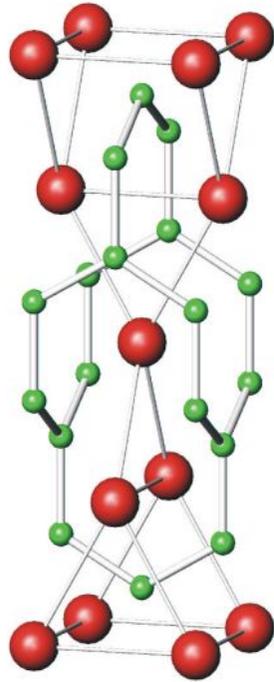
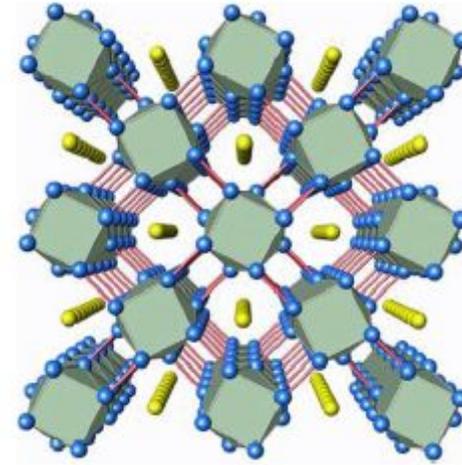
*oC52*

*oC16*

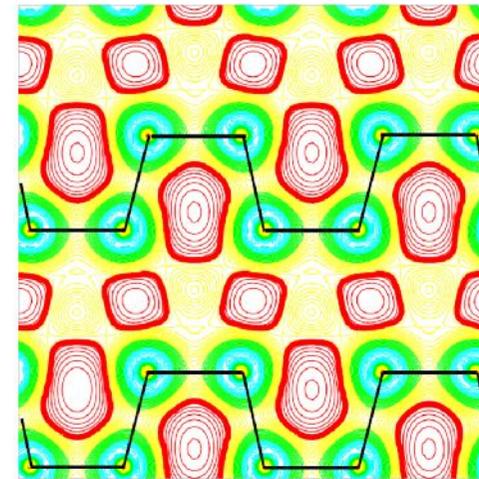


Incomm. host-guest

**Incommensurate** host-guest structure Rb-IV.  
Positions of host atoms (blue circles)  
correspond to the W-cation sublattice of  
 $W_5Si_3$ . Yellow circles are guest atoms



Cs-IV structure. Arrangement of Cs atoms  
(red circles) resembles the Th-cation sublattice  
of  $ThSi_2$ . **The electron clouds (green circles)**  
**'imitate' absent Si anions**



Valence electron density in **Li-cI16**.  
Black lines connect the neighbor atoms.  
Red maxima in the interstitials  
**'mimic' absent As anions** in  $Yb_4As_3$

## $p - T$ phase diagrams: Quasiharmonic approach

$$E_{\text{tot}}\{n(\mathbf{r})\} = T_0\{n(\mathbf{r})\} + \frac{e^2}{2} \int \frac{n(\mathbf{r}) n(\mathbf{r}')}{|\mathbf{r} - \mathbf{r}'|} d\mathbf{r} d\mathbf{r}' +$$

$T = 0:$

$$+ \int n(\mathbf{r}) V_{\text{ext}}(\mathbf{r}) d\mathbf{r} + E_{\text{xc}}\{n(\mathbf{r})\} + \frac{e^2}{2} \sum_{N, N'} \frac{Z_N Z_{N'}}{|\mathbf{R}_N - \mathbf{R}_{N'}|}$$

$$\left| p(V) = -\frac{\partial E_{\text{tot}}}{\partial V} \quad \left| H = E_{\text{tot}} + pV \right. \right.$$

$0 < T < E_F:$

$$\left| G = F + pV \right.$$

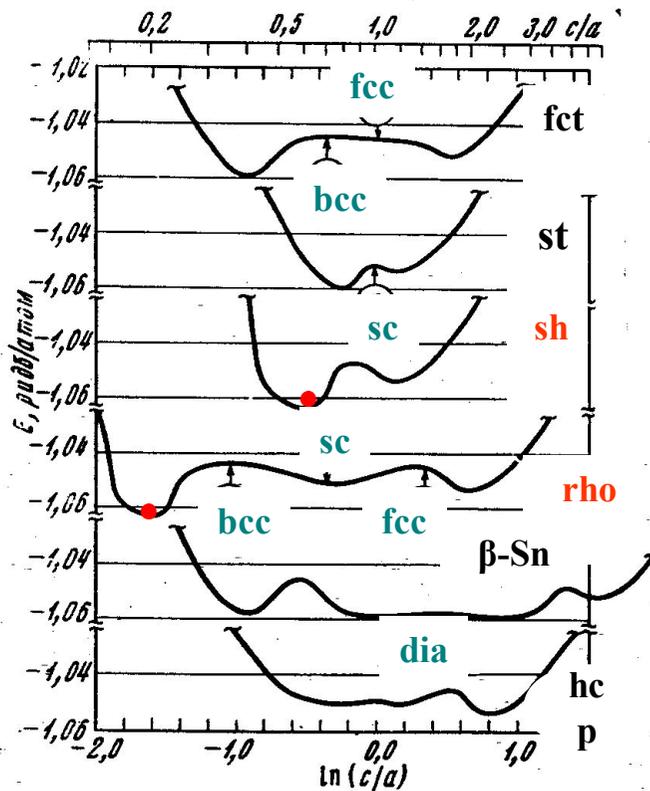
$$F = E_{\text{tot}}(V) + F_{\text{ph}}(V, T)$$

$$F(V, T) = E(V) + \frac{1}{2} \sum_{\mathbf{q}, \lambda} \hbar \omega_{\mathbf{q}\lambda}(V) - k_B T \sum_{\mathbf{q}, \lambda} \ln \frac{1}{1 - \exp(-\hbar \omega_{\mathbf{q}\lambda}(V) / k_B T)}$$

# Metallic hydrogen [Brovman, Kagan & Holas *JETP* 1971]

$$E_{cr} = E_{el} + \frac{1}{2} \sum_{n,n'} \frac{e^2}{|\mathbf{R}_n - \mathbf{R}_{n'}|} + E_{ZPM} \quad E_{ZPM} = \frac{1}{2} \sum_{\mathbf{q},\lambda} \hbar \omega_{\mathbf{q}\lambda}$$

Electron energy  $E_{el} = E^{(0)} + E^{(2)} + E^{(3)} + E^{(4)} + \dots$  is expanded as a series in the electron-proton coupling  $V_G/E_F$ , where  $V_G = \frac{4\pi e^2}{G^2 \Omega_{at}}$



sh, rho + 5 triclinic structures

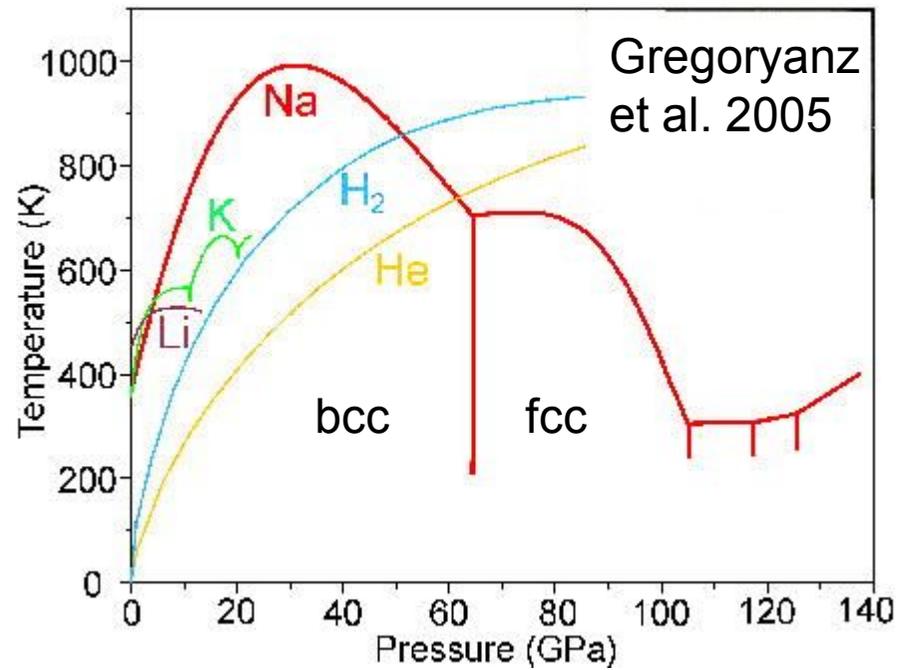
- All cubic structures are not favored energetically
- fcc, bcc, and hcp structures are dynamically unstable up to  $\sim 20$  Mbar ( $r_s \sim 1.2$ )
- Min. sh + min. rho + 5 triclinic structures form a unified family of structures
- At  $1 \leq r_s \leq 1.65$ , or 1 - 100 Mbar, highly anisotropic (chain-like and layered) structures are energetically favorable

EG Maksimov & DY Savrasov

*Solid State Commun.* 2001

$T_c = 600$  K

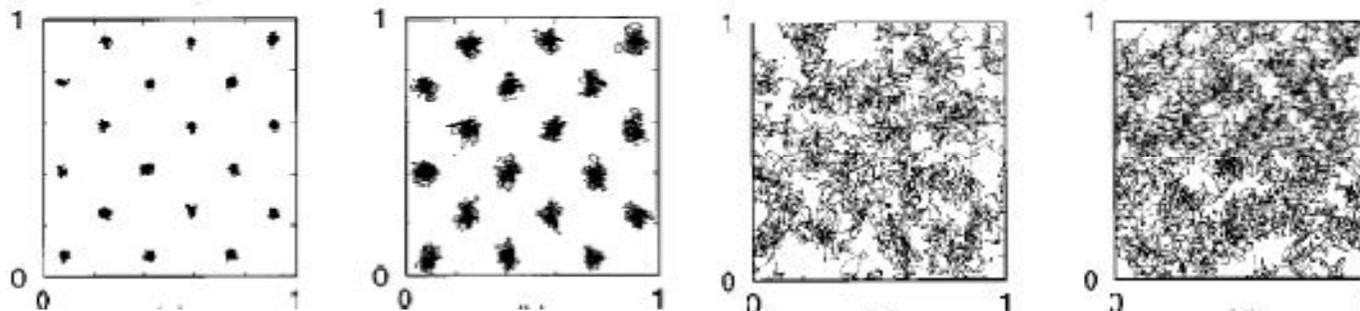
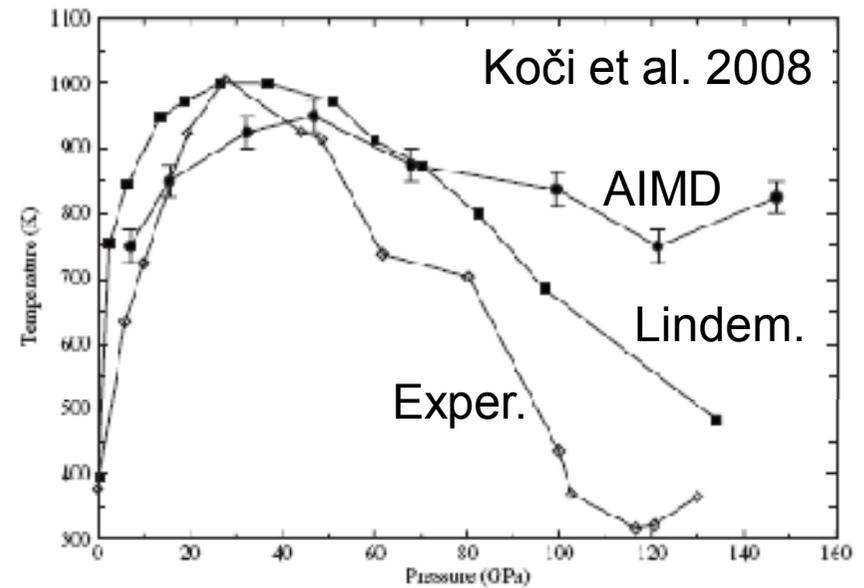
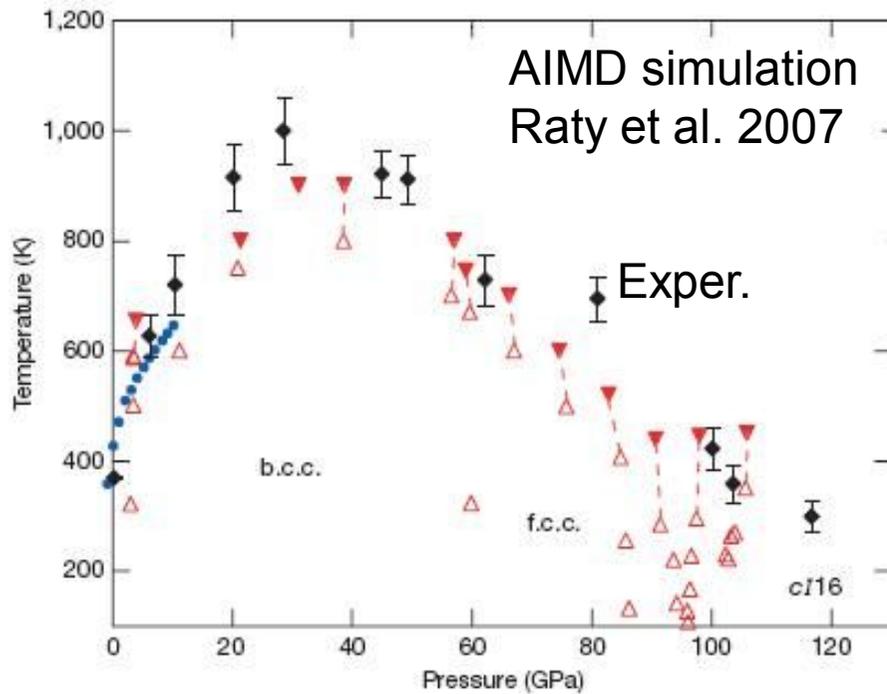
# Anomalous melting of Na



## Clausius–Clapeyron equation

$$\frac{dT_m}{dp} = \frac{\Delta V}{\Delta S}$$

# Melting of Na: *ab initio* simulations

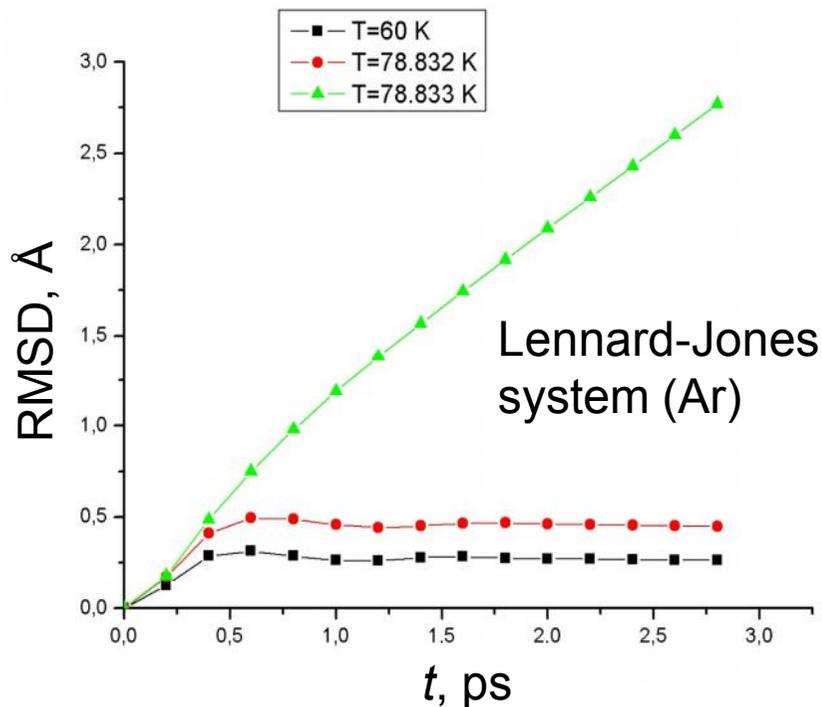


Atomic displacements as a function of temperature

Lindemann's criterion:  $\sqrt{\langle u^2(T_m) \rangle} = Ld_{\text{NN}}$

where  $\langle u^2(T) \rangle = \frac{\hbar}{2MN} \sum_{\mathbf{q}, \lambda} \frac{\coth(\hbar\omega_{\mathbf{q}\lambda} / 2k_{\text{B}}T)}{\omega_{\mathbf{q}\lambda}}$

$g(\omega) = \frac{1}{N} \sum_{\mathbf{q}, \lambda} \delta(\omega - \omega_{\mathbf{q}\lambda})$        $\langle u^2(T) \rangle = \frac{\hbar}{2M} \int_0^\infty \frac{d\omega}{\omega} g(\omega) \coth \frac{\hbar\omega}{2k_{\text{B}}T}$



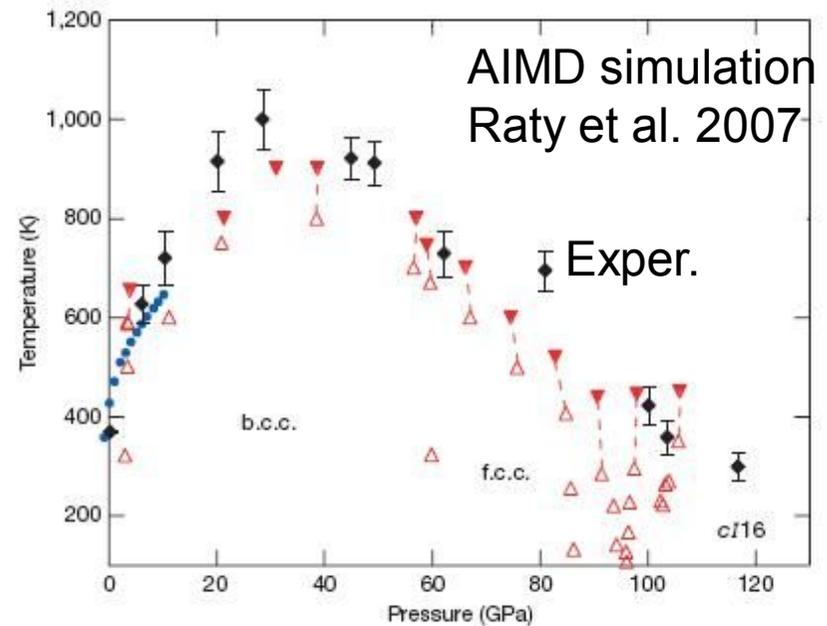
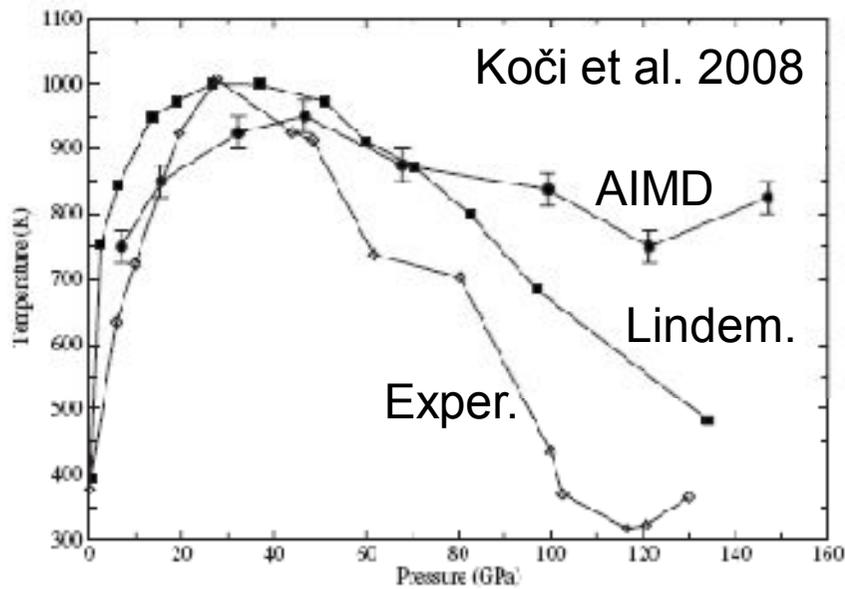
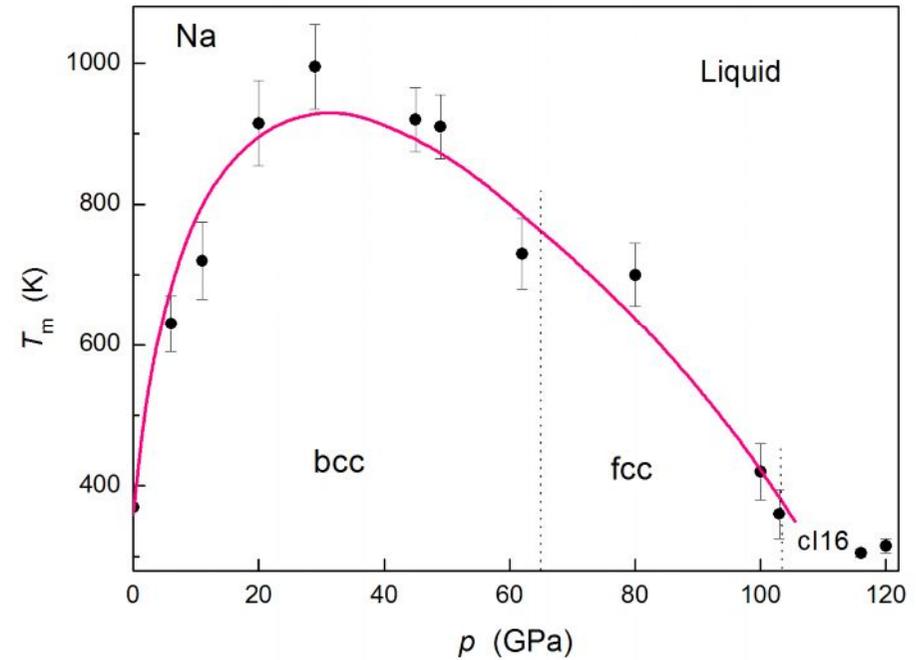
at high  $T$

$$T_m = \frac{(Ld_{\text{NN}})^2}{k_{\text{B}}} \frac{M}{\langle \omega^{-2} \rangle}$$

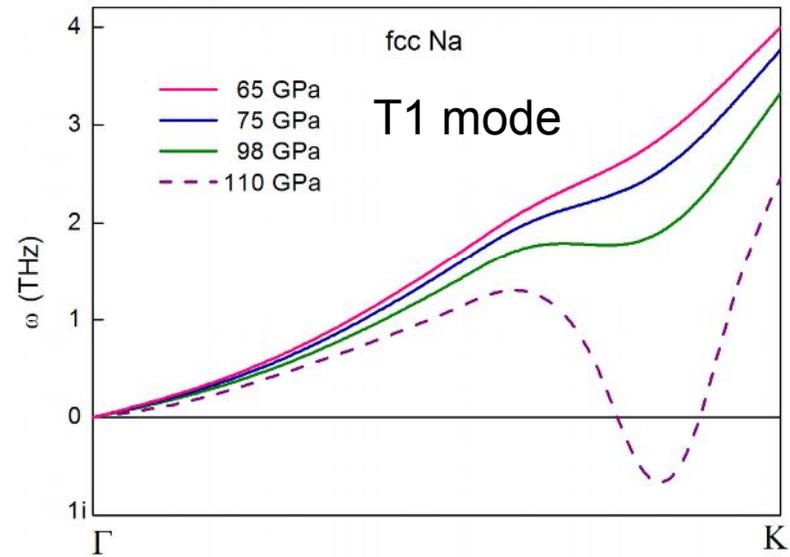
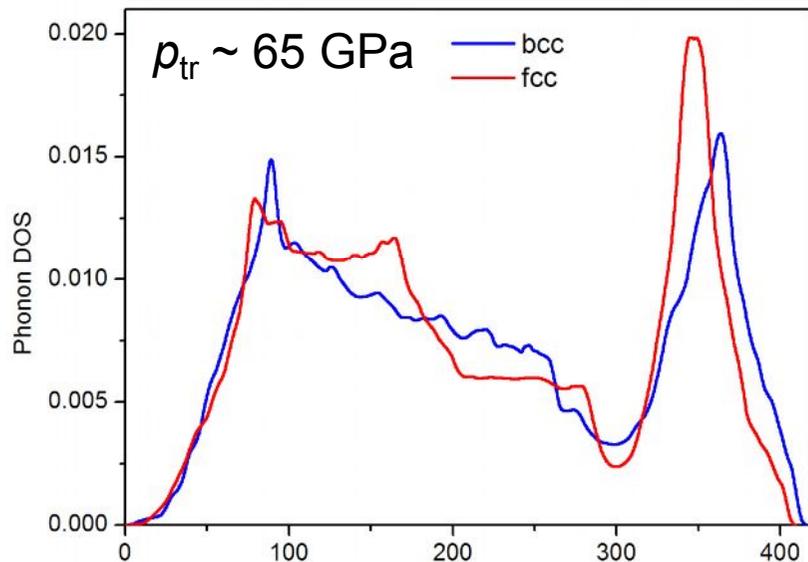
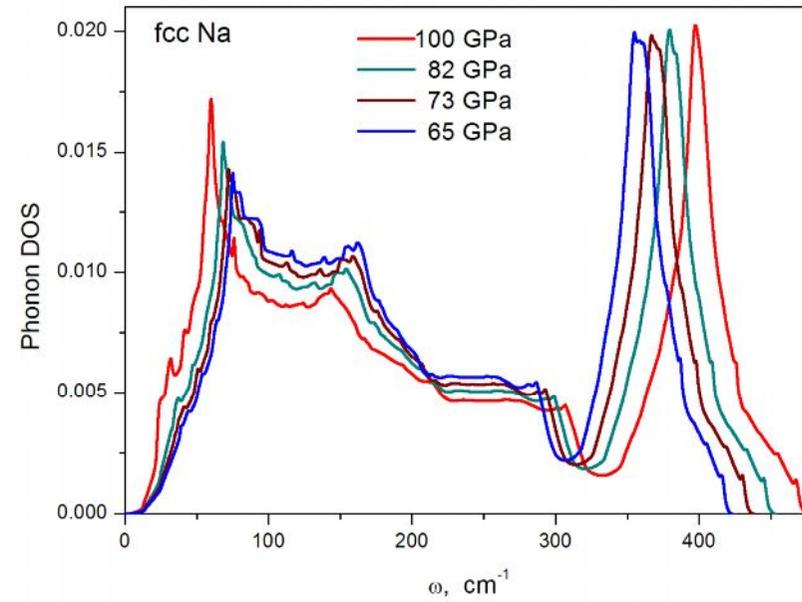
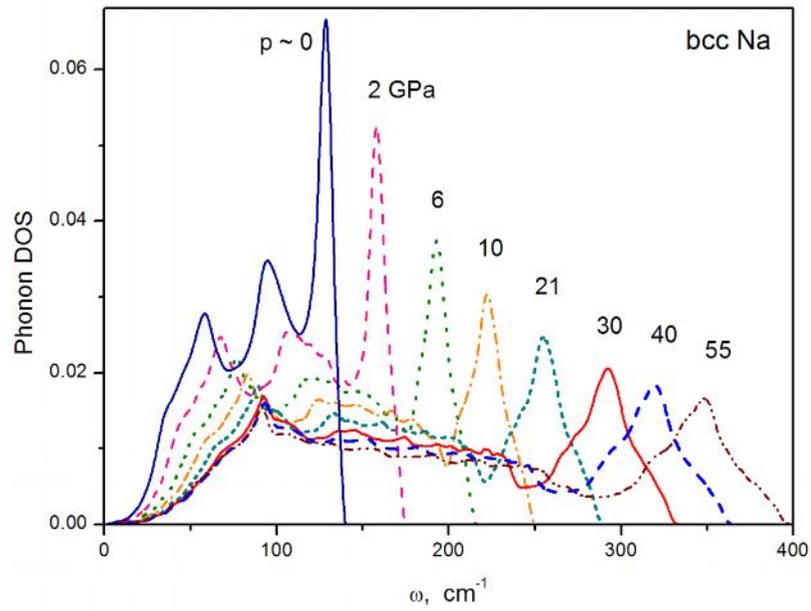
$$\langle \omega^{-2} \rangle = \int_0^\infty \frac{g(\omega)d\omega}{\omega^2}$$

# Melting curve of Na

*Lepeshkin et al. JETP Lett. 2009*

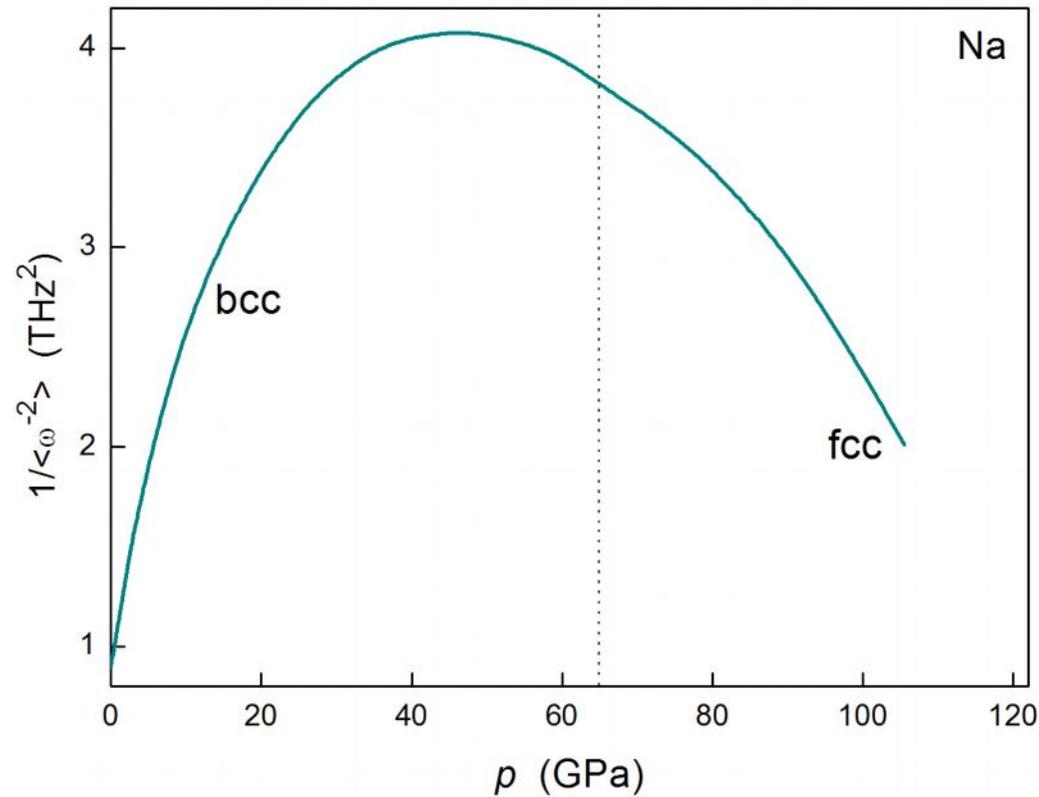


# Na: phonon spectra at various pressures



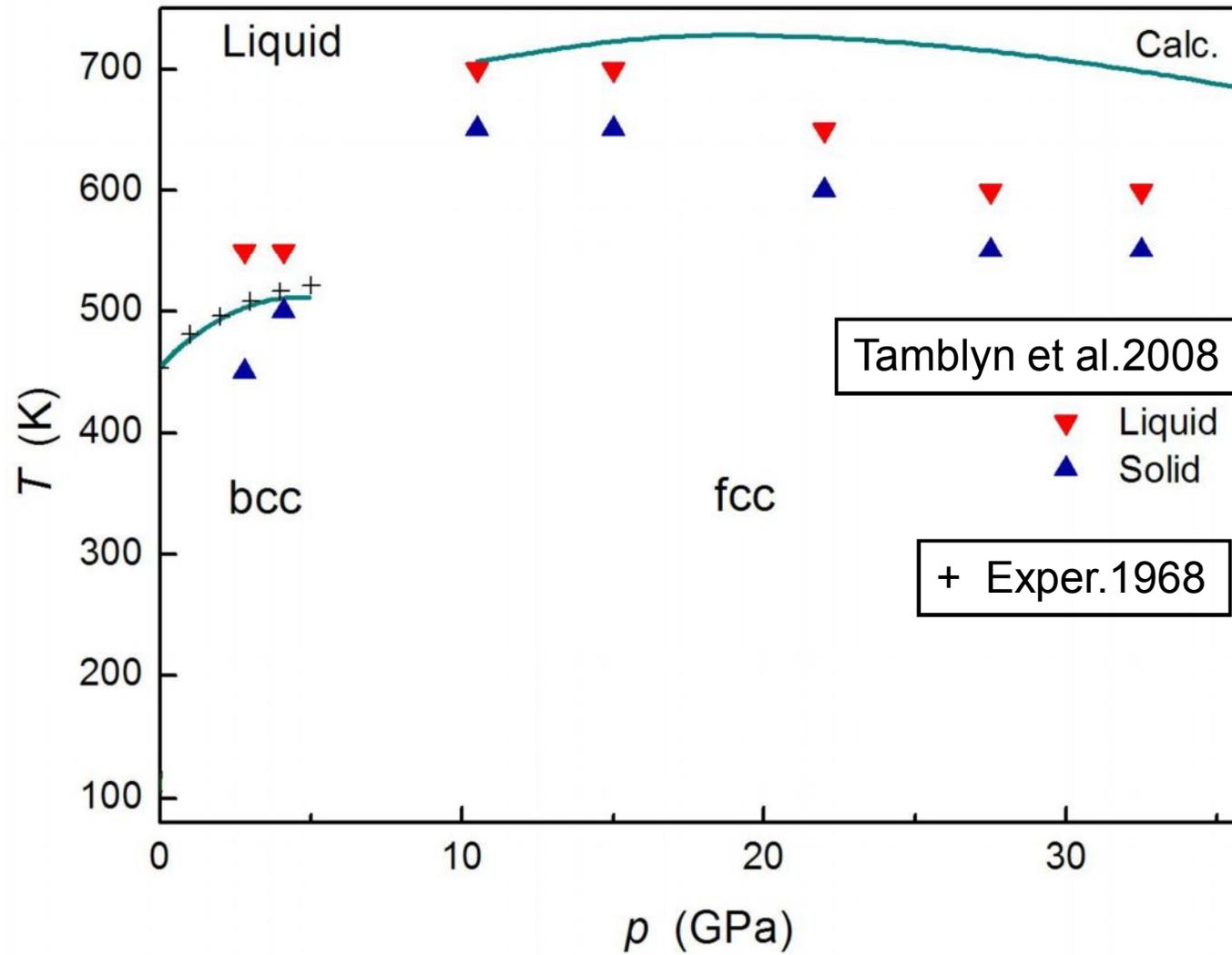
# Characteristic frequency of Na

$$T_m \approx \frac{(Ld_{NN})^2}{k_B} \frac{M}{\langle \omega^{-2} \rangle} \quad \langle \omega^{-2} \rangle = \int_0^{\infty} \frac{g(\omega)d\omega}{\omega^2}$$

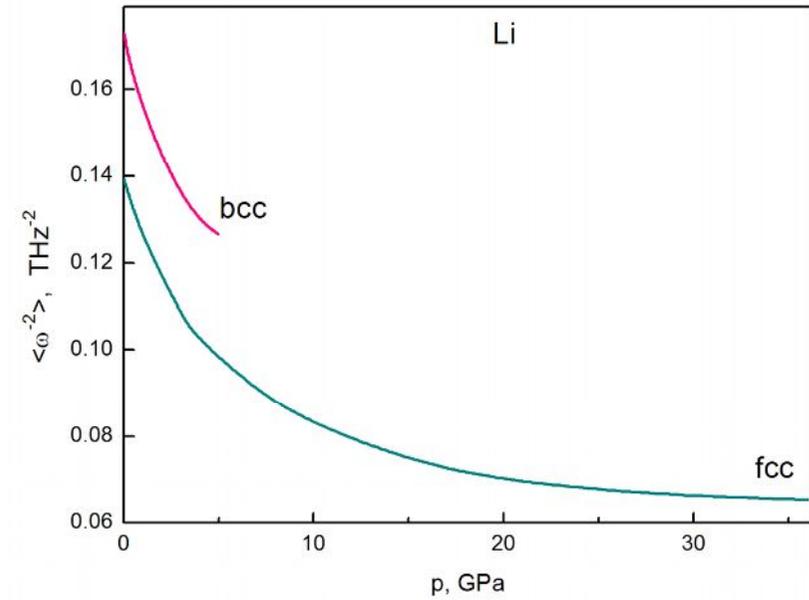
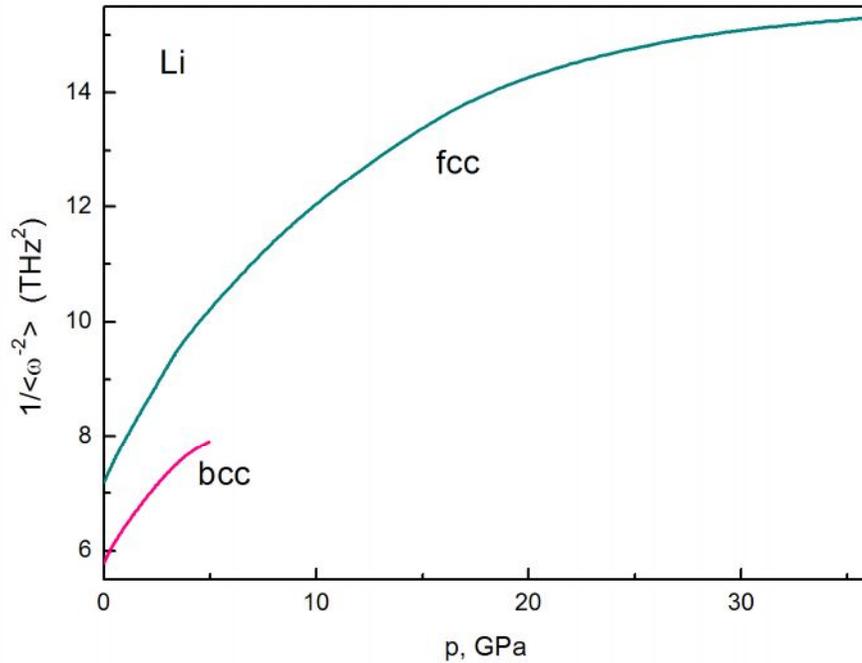


At  $p_{tr} = 65$  GPa,  $\langle \omega^{-2} \rangle_{bcc} = \langle \omega^{-2} \rangle_{fcc}$

# Melting curve of Li [Lepeshkin et al. JETP Lett. 2009]



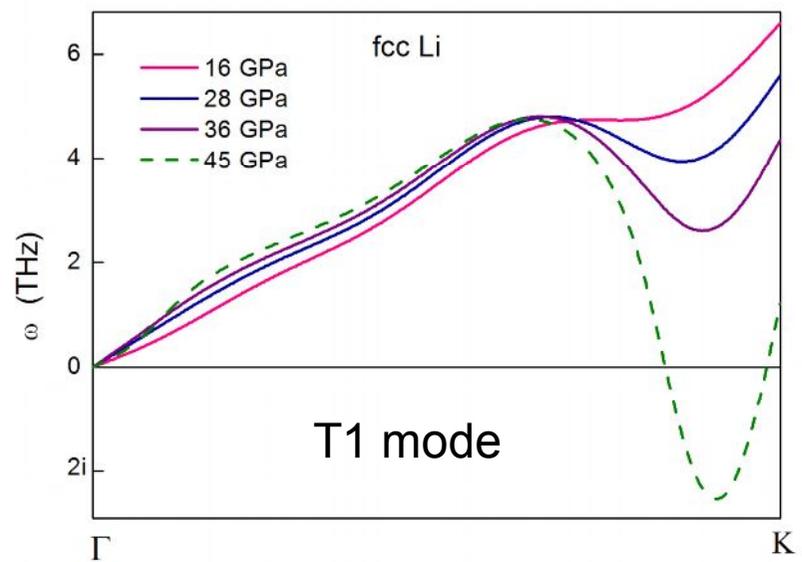
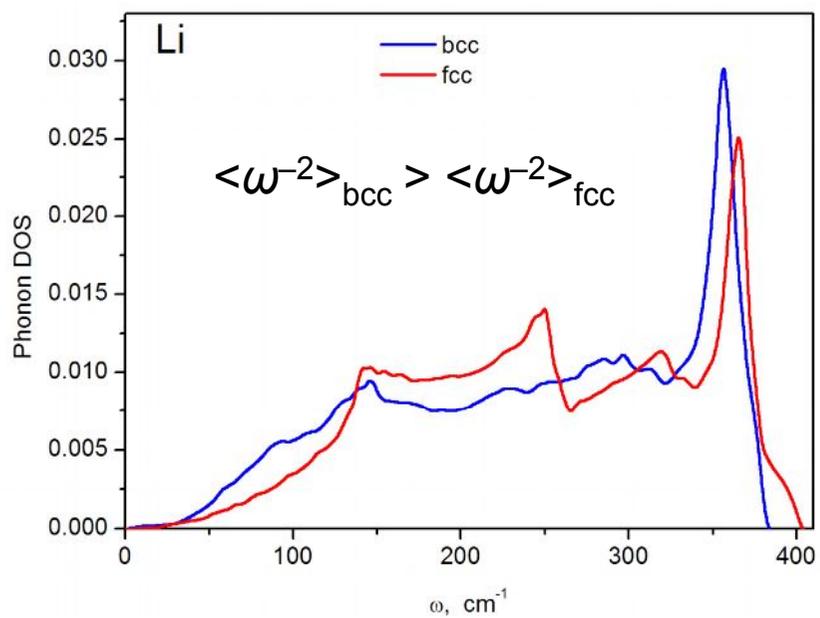
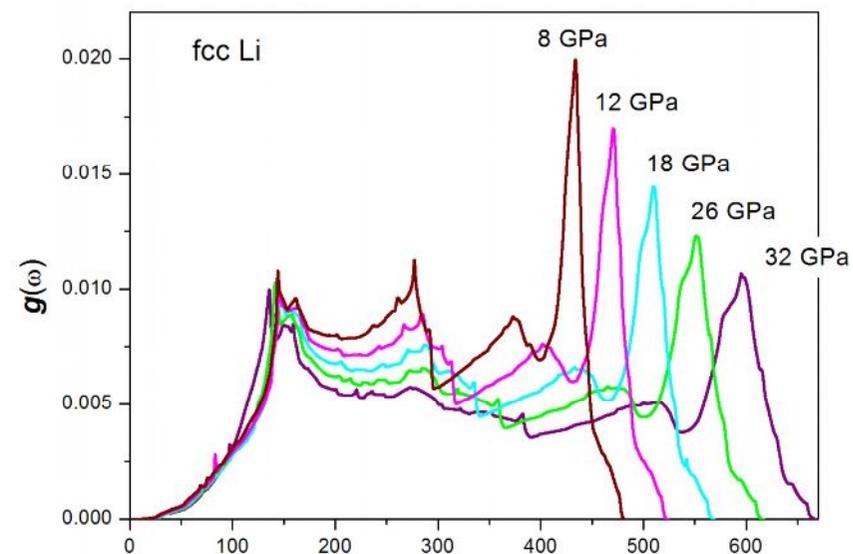
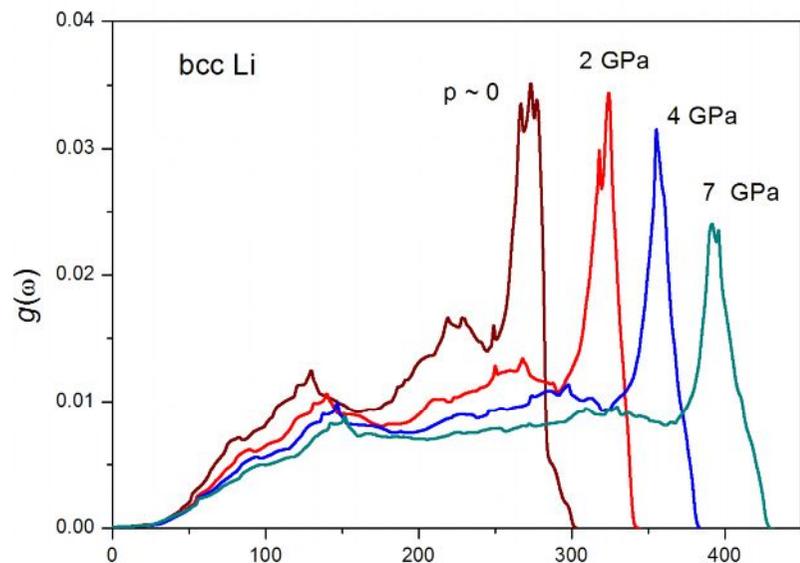
# Lithium: Characteristic frequency $\langle \omega^{-2} \rangle$



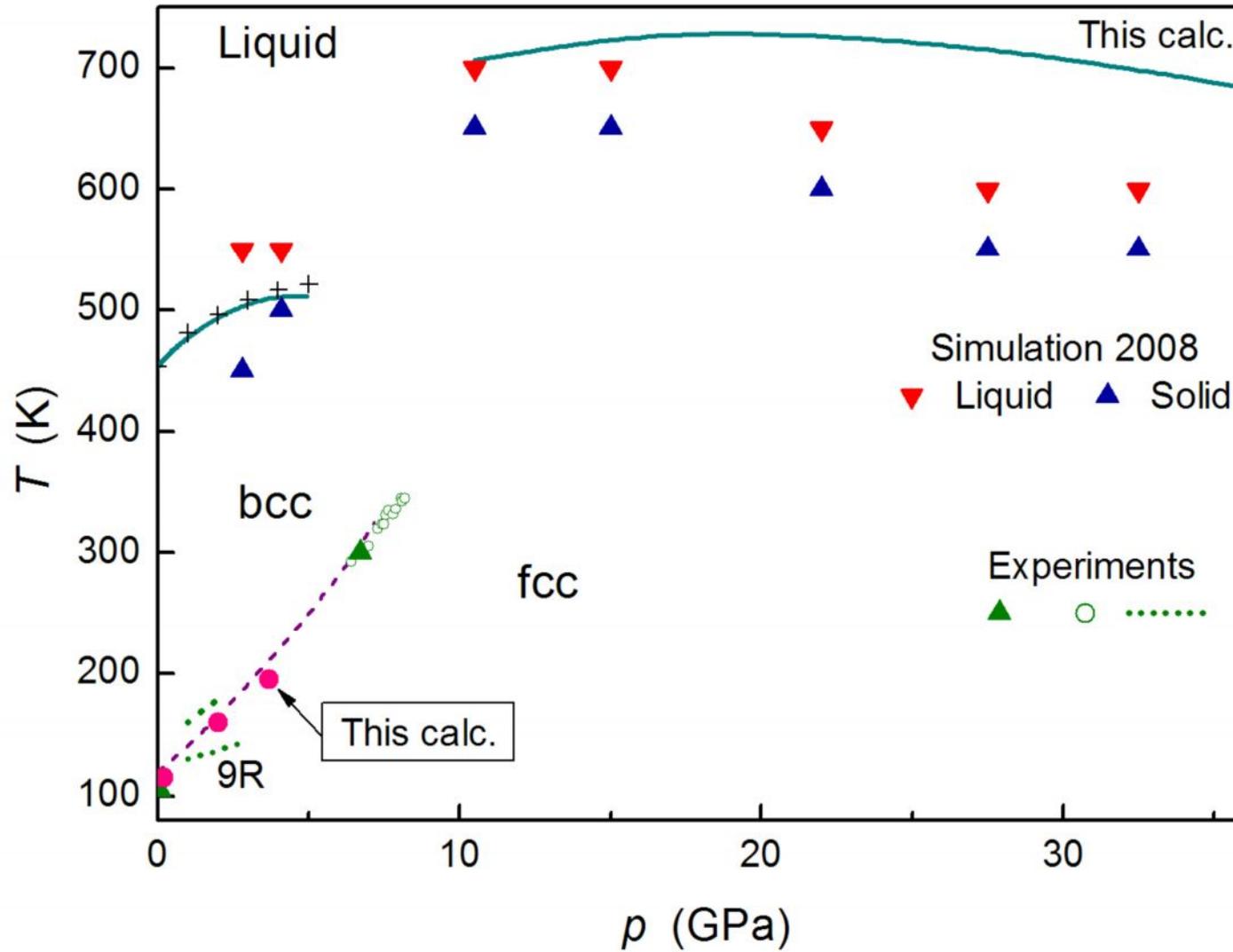
At any pressure, there is a jump in the characteristic frequency:

$$\frac{1}{\langle \omega^{-2} \rangle_{\text{bcc}}} < \frac{1}{\langle \omega^{-2} \rangle_{\text{fcc}}}$$

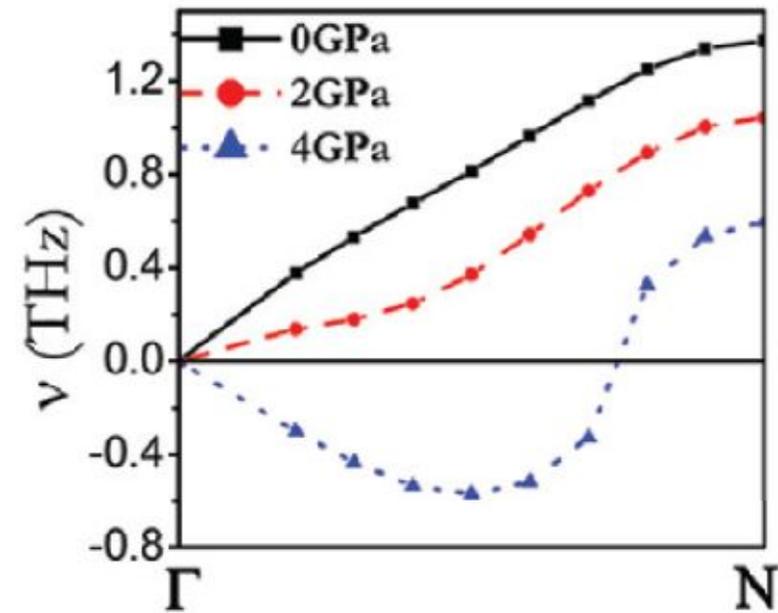
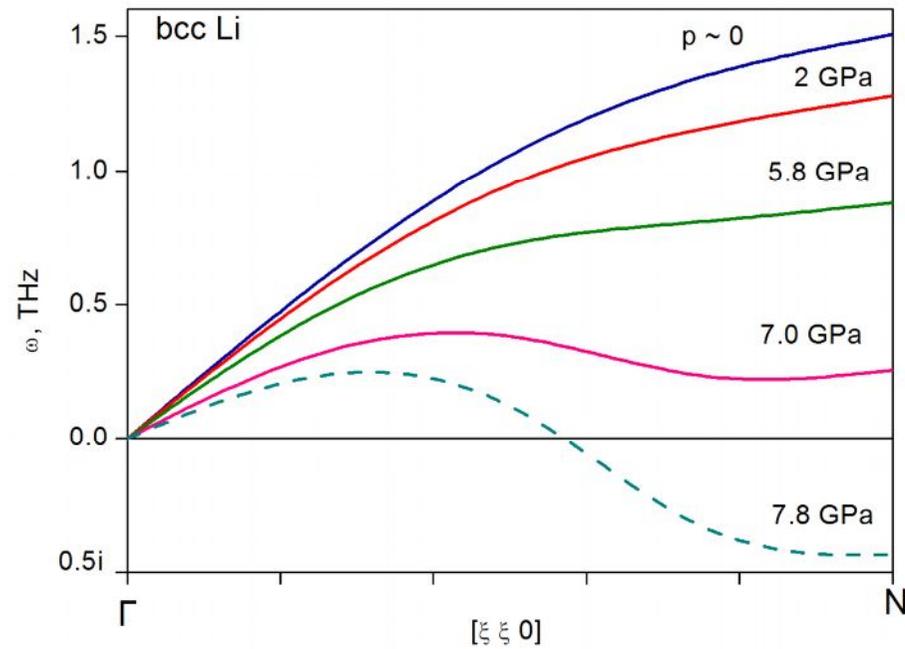
# Phonon spectra of Li at various $p$



# Phase diagram of Li

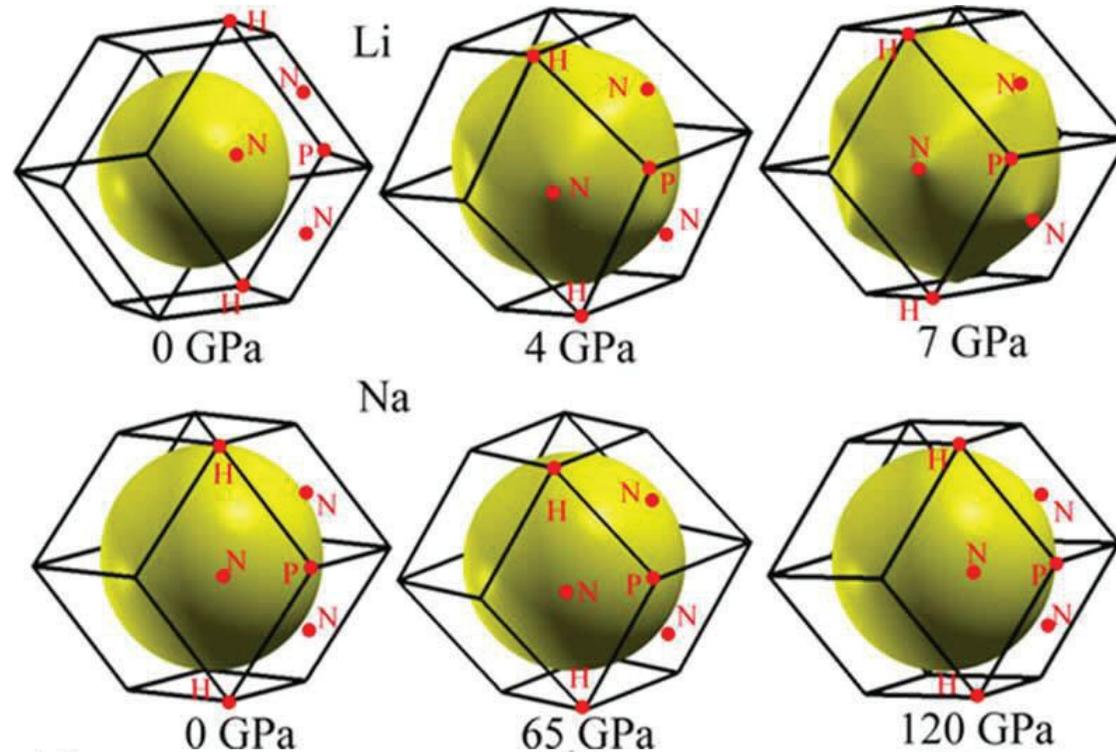


# Phonon dispersions of bcc Li (transversal T1 mode)

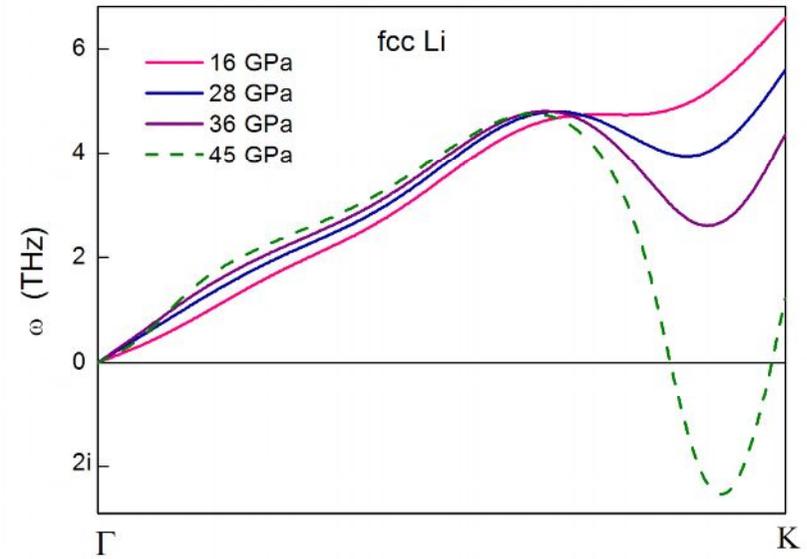
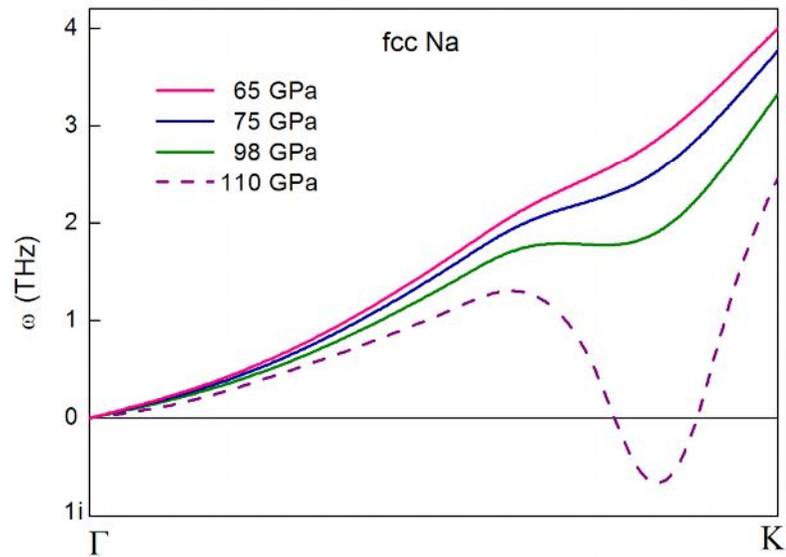


*Xie et al. 2008*

# Fermi surfaces of bcc Li and Na at various pressures



# fcc Na and Li: dispersions of T1 mode at various $p$

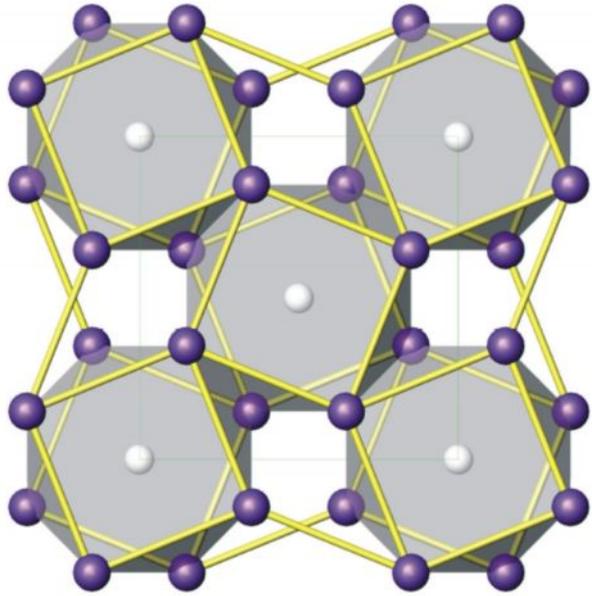


Phonon instabilities of different types



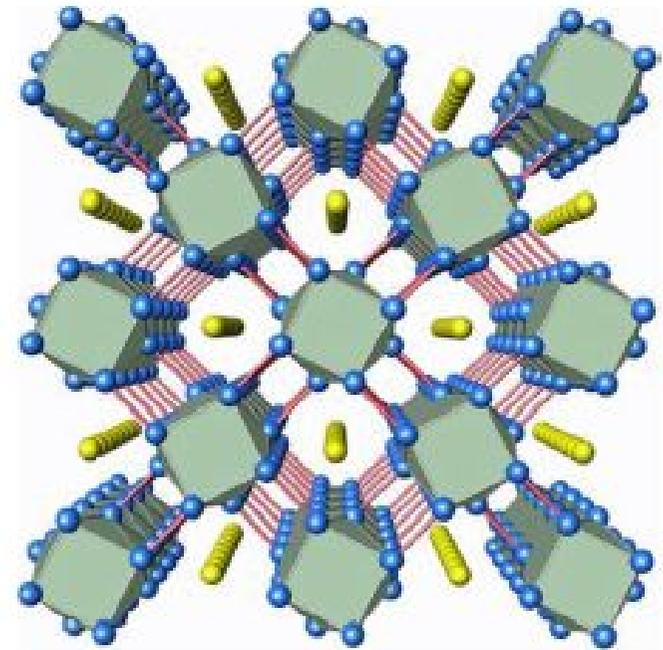
# Conclusions

## Composite (host-guest) structures

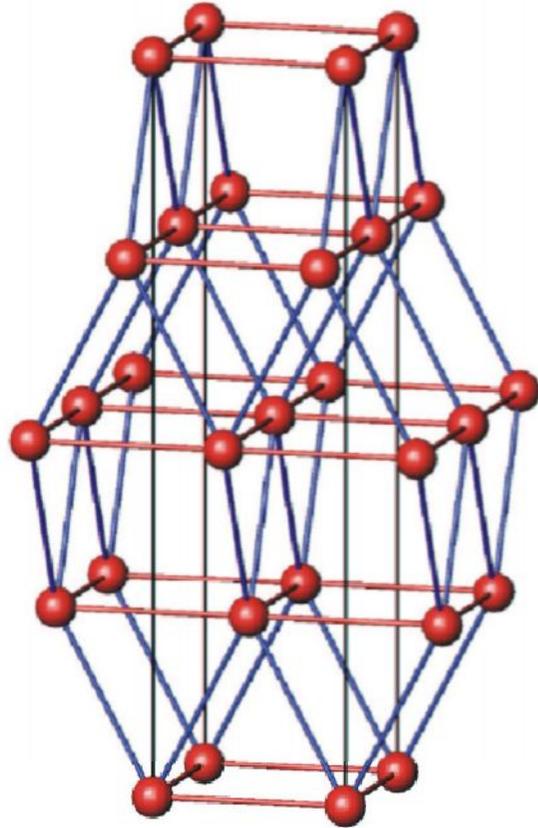


Crystal structure of the Ba-IV phase along the tetragonal  $c$ -axis. One set of Ba atoms forms a framework ( $Z=8$ ) which is filled with *incommensurate* chains of a second set of Ba atoms. This type of structure was later found for Sr and group-V elements (As, Sb, Bi) [R. Nelmes et al. PRL 1999]

*Incommensurate* crystal structure of the Rb-IV phase. The arrangement of guest atoms is similar to the cation sublattice of  $W_5Si_3$  structure. A similar structure was also found for K-III high-pressure phase. [U. Schwarz et al. PRL 1999]



## Cs-IV structure



The bct crystal structure of Cs-IV ( $Z = 4$ ). The structure is made up of layers of collinear face-sharing trigonal prisms, the layers being stacked along the vertical  $c$ -axis. The prism orientation changes by  $90^\circ$  in subsequent layers

[*K. Takemura et al. PRL 1982*]

The  $\text{ThSi}_2$  structure. The cation (Th) sublattice resembles the arrangement of Cs atoms in the Cs-IV structure. The electrons in Cs-IV ‘imitate’ absent anions in the Si sites

